

## Synthesis and structure of ferrocenylphosphinic acids



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### ABSTRACT

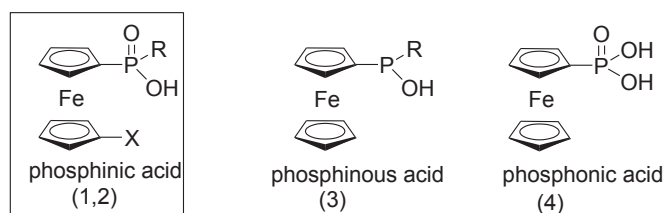
The series of ferrocenylphosphinic acids  $\text{Fc}(\text{P}(\text{R})(\text{O})(\text{OH}))$  (**1a–d**) and ferrocene-1,1'-diyl-phosphinic acids  $\text{Fc}(\text{P}(\text{R})(\text{O})(\text{OH}))_2$  (**2a–d**) ( $\text{R} = \text{H}$  (a), Me (b), Et (c), Ph (d)) have been obtained and studied using X-ray single crystal diffraction, IR-spectroscopy and quantum chemistry. The intermolecular hydrogen bonding in **1a–d** proceeds highly stereoselectively and results in formation of racemic conglomerate of **1d**. The formation of inter- and intra-molecular hydrogen bonds for disubstituted acids **2c–d** depends on substituent at phosphorus atoms.

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### Introduction

The arylphosphinic acids  $\text{ArP}(\text{R})(\text{O})(\text{OH})$  ( $\text{R} = \text{H}$ , alkyl, aryl) are widely used for the construction of self-assembled two- and three-dimensional structures or single molecular magnets (SMM) based on p- [1], d- [2] and f- [3] metals, e.g., Ru [4] and Mn [5] complexes. The replacement of the aryl fragment in these acids for the cognate ferrocenyl fragment, containing redox-active iron atom, may result in additional interesting properties like enhanced exchange interactions and give the opportunity to make sensors or switchable devices.

However ferrocenylphosphinic acids (**1,2**) are poorly studied in contrast to other ferrocenyl-containing phosphorus acids. For example, ferrocenylphosphinous acids (**3**) existing as the mixtures of two tautomeric forms have been widely employed as ligands in transition metals catalysis of cross-coupling reactions [6,7], asymmetric hydrogenation [8] etc. Ferrocenylphosphonic acid (**4**) has been considered as redox-active pH-responsive molecule and the redox half-wave potential of ferrocenylphosphonates is sensitive to the bonding mode with metal atoms [9–14].



$\text{R} = \text{H}$ :  $\text{X} = \text{H}$  (**1a**),  $\text{P}(\text{O})(\text{OH})\text{H}$  (**2a**)  
 $\text{R} = \text{Me}$ :  $\text{X} = \text{H}$  (**1b**),  $\text{P}(\text{O})(\text{OH})\text{Me}$  (**2b**)  
 $\text{R} = \text{Et}$ :  $\text{X} = \text{H}$  (**1c**),  $\text{P}(\text{O})(\text{OH})\text{Et}$  (**2c**)  
 $\text{R} = \text{Ph}$ :  $\text{X} = \text{H}$  (**1d**),  $\text{P}(\text{O})(\text{OH})\text{Ph}$  (**2d**)

The data on the synthesis, structure and redox properties of ferrocenylphosphinic acids are scarcely presented, although their first synthesis dates back to 1962: Sollott and Howard found that ferrocene reacts with  $\text{PCl}_3$  under Fridel–Crafts reaction conditions in the presence of  $\text{AlCl}_3$  to form (**1a**) and (**2a**) with the yield of 3.2% and 4.6% respectively after the hydrolysis [15]. Diferrocenylphosphinic acid was also isolated from this mixture [16]. The yield of (**1a**) was slightly improved to 8.6% using  $\text{Me}_2\text{NPCl}_2$  as starting reagent instead of  $\text{PCl}_3$  [17]. The formation of (**1a**) as a side-product has been also observed in the hydrolysis of diphosphene

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